Crystallographic report

catena-Poly[aqua(phenyl-1,3-bisoxyacetato) tris(pyridine)cadmium(II)], [(H₂O)(C₅H₅N)₃ (O₂CCH₂OC₆H₄OCH₂CO₂)Cd]_n

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The $[O_2CCH_2OC_6H_4OCH_2CO_2]^{2-}$ group bridges adjacent planar $[(H_2O)(C_5H_5N)_3Cd]^{2+}$ units into a linear chain. The carboxyl groups occupy opposite sites in the octahedral environment of the cadmium atom. Copyright © 2004 John Wiley & Sons, Ltd.

KEYWORDS: crystal structure; cadmium; phenyl-1,3-bisoxyacetatic acid; polymer structure

COMMENT

In recent years, self-assembly metal complexes with multifunctional carboxylic acids ligand have attracted considerable attention in supramolecular chemistry and crystal engineering. The reaction of pyridine, cadmium dinitrate tetrahydrate and phenyl-1,3-bisoxyacetate sodium afforded the cadmium(II) coordination polymer $[(H_2O)(C_5H_5N)_3(O_2CCH_2OC_6H_4OCH_2CO_2)Cd]_n$ (Fig. 1). Crystallography shows that the cadmium atom is coordinated by two oxygen atoms of two phenyl-1,3-bisoxyacetate ligand, three pyridine nitrogen atoms and one water molecule that define an octahedral geometry. The $[O_2CCH_2OC_6H_4OCH_2CO_2]^{2-}$ ligand bridges adjacent planar $[(H_2O)(C_5H_5N)_3Cd]^{2+}$ units into a linear chain that propagates along the b axis. The two symmetry-independent chains have similar coordination geometries and topologies.

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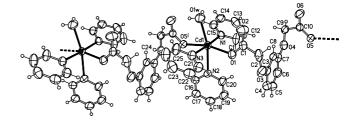


Figure 1. A portion one of the two polymeric $[(H_2O)(C_5H_5N)_3]$ $(O_2CCH_2OC_6H_4OCH_2CO_2)Cd]_n$ chains. Selected bond distances and angles for the unprimed molecule: Cd1-O1 2.261(5), Cd1-O5 i 2.342(4), Cd1-O1w 2.354(5), Cd1-N1 2.332(5), Cd1-N2 2.475(5), Cd1-N3 2.333(6) Å; O1-Cd1-O5 i 163.4(2), O1-Cd1-O1w 113.2(2), O1-Cd1-N1 92.3(2), O1-Cd1-N2 81.5(2), O1-Cd1-N3 90.7(2), O5 i -Cd1-O1w 83.4(2), O5 i -Cd1-N1 88.3(2), O5 i -Cd1-N2 81.9(2), O1w-Cd1-N2 89.2(2), O1w-Cd1-N2 165.0(2), O1w-Cd1-N3 88.1(2), N1-Cd1-N2 92.8(2), N1-Cd1-N3 176.6(2), N2-Cd1-N3 89.2(2) $^\circ$. Translation code: i=x,y-1,z.

EXPERIMENTAL

The compound was synthesized by the addition of pyridine (3 ml) and cadmium dinitrate tetrahydrate (6.16 g, 20 mmol) to an aqueous solution of phenyl-1,3-bisoxyacetatic acid (9.04 g, 40 mmol). Colorless prisms were isolated from the filtered solution at room temperature over several days. Anal. Found: C, 50.89; H, 4.37; N, 7.01. Calc. for $C_{25}H_{25}N_3O_7Cd$: C, 50.73; H, 4.26; N, 7.10%. The

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room-temperature X-ray intensities were measured on a Rigaku Raxis-Rapid diffractometer for a colorless $0.18\times0.26\times0.38~\mathrm{mm}^3$ crystal. $C_{25}H_{25}CdN_3O_7$, M=591.88, triclinic, $P\overline{1}$, a=8.699(2), b=10.322(2), c=29.362(6) Å, $\alpha=80.17(3)$, $\beta=87.28(3)$, $\gamma=81.95(3)^\circ$, V=2571.4(9) Å 3 , Z=4, 11460 unique reflections, R=0.080 for 9392 $I>2\sigma(I)$ reflections. The final difference Fourier map had two large peaks (>4 e^ Å^{-3}), one at 1.35 Å from the C8 atom and the other at the same distance from the C8' atom, but is otherwise featureless. The spurious peaks are related to the non-crystallographic relationship between the two independent molecules (the unprimed atoms being related to the primed atoms by $(\frac{1}{2}+x,\frac{1}{2}-z)$, but the treatment of such pseudosymmetry was not attempted. Programs

used: SHELXS-97, SHELXL-97, ORTEP-II. CCDC deposition number: 235936

REFERENCES

- 1. Yuan W, Long LS. Appl. Organometal. Chem. 2003; 17: 257.
- Kumagai H, Inoue K, Kurmoo M. Bull. Chem. Soc. Jpn. 2002; 75: 1283.
- 3. Gao S, Liu JW, Huo LH, Ng SW. Acta Crystallogr. Sect. E 2004; 60: m462.